Chemistry Central Journal

RESEARCH ARTICLE

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Validation of an analytical method based on the high-resolution continuum source flame atomic absorption spectrometry for the fast-sequential determination of several hazardous/priority hazardous metals in soil

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Abstract

Background: The aim of this paper was the validation of a new analytical method based on the high-resolution continuum source flame atomic absorption spectrometry for the fast-sequential determination of several hazardous/priority hazardous metals (Ag, Cd, Co, Cr, Cu, Ni, Pb and Zn) in soil after microwave assisted digestion in aqua regia. Determinations were performed on the ContrAA 300 (Analytik Jena) air-acetylene flame spectrometer equipped with xenon short-arc lamp as a continuum radiation source for all elements, double monochromator consisting of a prism pre-monocromator and an echelle grating monochromator, and charge coupled device as detector. For validation a method-performance study was conducted involving the establishment of the analytical performance of the new method (limits of detection and quantification, precision and accuracy). Moreover, the Bland and Altman statistical method was used in analyzing the agreement between the proposed assay and inductively coupled plasma optical emission spectrometry as standardized method for the multielemental determination in soil.

Results: The limits of detection in soil sample (3σ criterion) in the high-resolution continuum source flame atomic absorption spectrometry method were (mg/kg): 0.18 (Ag), 0.14 (Cd), 0.36 (Co), 0.25 (Cr), 0.09 (Cu), 1.0 (Ni), 1.4 (Pb) and 0.18 (Zn), close to those in inductively coupled plasma optical emission spectrometry: 0.12 (Ag), 0.05 (Cd), 0.15 (Co), 1.4 (Cr), 0.15 (Cu), 2.5 (Ni), 2.5 (Pb) and 0.04 (Zn). Accuracy was checked by analyzing 4 certified reference materials and a good agreement for 95% confidence interval was found in both methods, with recoveries in the range of 94–106% in atomic absorption and 97–103% in optical emission. Repeatability found by analyzing real soil samples was in the range 1.6–5.2% in atomic absorption, similar with that of 1.9–6.1% in optical emission spectrometry. The Bland and Altman method showed no statistical significant difference between the two spectrometric methods for 95% confidence interval.

Conclusions: High-resolution continuum source flame atomic absorption spectrometry can be successfully used for the rapid, multielemental determination of hazardous/priority hazardous metals in soil with similar analytical performances to those in inductively coupled plasma optical emission spectrometry.

Keywords: High-resolution continuum source flame atomic absorption spectrometry, Inductively coupled plasma optical emission spectrometry, Bland and Altman statistical method, Soil analysis, Hazardous/priority hazardous metal

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Background

Twenty five years ago low-resolution line source atomic absorption spectrometry (LR-LS AAS), although in full maturity and providing high sensitivity and selectivity through the element-specific line radiation source, had limited versatility and low speed of analysis being known as a single-element technique. To face the challenges of the inductively coupled plasma optical emission spectrometry (ICP-OES), which imposed itself as a fast multielemental technique due to the charged-coupled device (CCD) technology, the need to increase the versatility of AAS with respect to line selection, background correction and higher analysis speed became imperative. Thus, in the mid-1990s a research group led by Becker-Ross from the Institute for Analytical Sciences, Berlin, Germany, designed and built the first instrument for high-resolution continuous source atomic absorption spectrometry (HR-CS AAS), in which all components were optimized for the novel approach [1-3]. The new concept instrument has been for several years commercially available for both flame (HR-CS FAAS) and graphite furnace (HR-CS GFAAS) atomization. The relevant advantages of HR-CS AAS over LR-LS AAS are [4]: (i) use of just one source for all elements; (ii) visualization of the environment of the analytical line permitting superior background correction by simultaneous measurement of atomic and background absorption; (iii) fast-sequential multielemental determination using atomic lines or molecular bands (changing lines and parameters optimization only takes a few seconds); (iv) correction of the fine-structured background using a reference spectrum. The extremely well acceptance of HR-CS AAS instrument after its introduction on the market together with the versatility of the analytical applications was reflected in several reviews [5-11]. Both approaches of the technique, using either flame or electrothermal atomization, were used for multielemental determination in various complex matrices. Thus, HR-CS GFAAS was mentioned for elemental determination in food after acidic digestion or extraction in tetramethylammonium hydroxide [12-14], airborne particulates [15-18], biological (beans and grain) samples using direct solid sampling [19-21], fertilizer by slurry sampling [22], crude oil [23-26], biodiesel [27] and in whole blood directly in diluted/undiluted samples [28].

At the same time, HR-CS FAAS was used for multielemental determination in vegetable oil as microemulsion [29], plant tissues after mineralization [30,31], dairy products by slurry sampling [32], pewter alloys after digestion in HCl solution [33], lubricating oil after acid digestion, oil-in water emulsification and dilution in kerosene [34-36], and soil extracts [37].

The introduction of the continuum source and high-resolution monochromator has opened the door to monitor non-metals (P, S, F, Br, N, etc.) with sensitive lines in the far UV, not accessible for LS-AAS. Thus, the molecular absorption spectrometry (HR-CS MAS) can overcome this barrier by measuring the absorption of a line within the band spectra of a diatomic molecular radical (PO, CS, SH, AlF, SrF, GaF, AlBr, CaBr, NO, etc.) stable in flame or graphite furnace at the atomizer temperature [9,21,38-49].

This paper reports on the validation of an analytical method for the fast-sequential determination of several hazardous/priority hazardous metals (Ag, Cd, Cr, Co, Cu, Ni, Pb and Zn) in soil based on the HR-CS FAAS. The soil samples were subjected to microwave assisted digestion in aqua regia according to ISO 11466 – 1999. Validation was conducted according to the method-performance study to estimate the performance characteristics of the new method (limit of detection, limit of quantification, precision and accuracy [50,51]. Furthermore, the Bland and Altman statistical method was used in analyzing the agreement between the proposed assay and inductively coupled plasma optical emission spectrometry as standardized method for the multielemental determination in soil [52].

The paper is important for the analytical practice because currently the standardized methods for soil analysis by atomic absorption spectrometry are LR-LS FAAS and GFAAS [53]. On the other hand, ISO/IEC 17025–2005 and the Programme of Association of Official Analytical Chemistry demand the fit-for-purpose validation of the new techniques/assays before use.

Results and discussion

Limits of detection and quantification

The limits of detection and quantification found in HR-CS FAAS and ICP-OES according to the 3σ criterion are given in Table 1. The limit of detection ($3\sigma_b/m$) was computed considering the slope of the calibration curve (m) and the standard deviation of the background (σ_b) from 10 successive measurements of the signal corresponding to blank [50,51,54,55]. The blank was prepared in the same way as the soil samples using identical volumes of reagents. The limit of quantification corresponding to the lowest concentration being determined with a specified relative standard deviation, commonly 10%, was considered as 3 times the limit of detection. Based on the protocol of sample preparation, the limits of detection and quantification were expressed in dry mass (mg/kg). The calibration curves in HR-CS FAAS drawn on the range 0 – 1 µg/ml element had correlation coefficients of 0.9950 - 0.9999, close to those found in ICP-OES. The limits of quantification found in the analytical method based on HR-CS FAAS allow the determination of elements in soil at concentration levels below the limits of applicability given for LR-LS FAAS

Table 1 Limits of detection (3σ criterion) and quantification (mg/kg) in soil

	HR-CS-FAAS			ICP-OES			Limit of applicability of LR-LS-FAAS[c]
	λ (nm)	LOD[a]	LOQ[b]	λ (nm)	LOD[a]	LOQ[b]	
Ag	328.068	0.18	0.54	328.068	0.12	0.36	-
Cd	228.802	0.14	0.42	214.438	0.05	0.15	>2
Co	240.206	0.36	1.08	228.615	0.15	0.45	>12
Cr	357.869	0.25	0.75	267.716	1.4	4.2	>12
Cu	324.754	0.09	0.27	324.754	0.15	0.45	>5
Ni	232.003	1.0	3.0	305.082	2.5	7.5	>12
Pb	217.000	1.4	4.2	220.351	2.5	7.5	>15
Zn	213.856	0.18	0.54	213.856	0.04	0.12	>2

[a] – LOD calculated for an amount of 1.000 g soil mineralized and diluted to 100 ml solution.
[b] – LOQ considered three times LOD.
[c] – according to [53].

[53]. Data in Table 1 reveals better detection limits for Cr, Cu, Ni and Pb by HR-CS FAAS than by ICP-OES. A comparison of the detection limits found by us with those given in [52] for ICP-OES is not relevant, since the values are strongly depended on equipment and laboratory conditions. However, each laboratory should determine its specific LODs within a validation study.

Accuracy of the multielemental determinations

Accuracy of the analytical method based on HR-CS FAAS was assessed by analyzing 4 certified reference samples. The results in comparison with those found in ICP-OES are displayed in Table 2. The average recovery and pooled standard deviation of 95% confidence interval were computed based on the results obtained for each CRM. According to data in Table 2, the analytical method based on HR-CS FAAS provided good accuracy with recoveries in the range 94–106%, similar to the values of 97–103% found in ICP-OES. The good recovery of certified value in both methods indicated the absence of interferences. Moreover, the method of correction the continuous background and NO molecular absorption in HR-CS FAAS proved to be efficient in the case of the complex matrix of soil.

Precision in HR-CS FAAS and ICP-OES

The metal contents and standard deviations found in soil samples by HR-CS FAAS and ICP-OES are presented in Additional file 1. The pooled relative standard deviations in HR-CS FAAS were: 2.5% (Ag), 4.6% (Cd), 3.2% (Cr), 4.3% (Cu), 5.2% (Co), 3.2% (Ni), 2.2% (Pb), 1.6% (Zn), similar to those in ICP-OES: 4.7% (Ag), 3.2% (Cd), 2.3% (Cr), 6.1% (Cu), 4.4% (Co), 4.9% (Ni), 3.7% (Pb) and 1.9% (Zn). The values comply with the informative ones given in [52].

Statistical validation using the Bland and Altman method

The Bland and Altman plots are displayed in Figure 1 and Additional file 2, while the descriptive statistics for 95% confidence interval (CI) in Table 3.

The contents of 3 out of the 8 elements under study (Cu, Pb and Zn) covered a wide range of concentration with individual values grouped on specific sub-domains. Therefore, in the case of Cu, the samples were split in two groups, with contents below 1000 mg/kg (34 samples) and in the range 1000 – 22000 mg/kg (16 samples). The distribution of Pb contents imposed the grouping in three sub-groups: below 1000 mg/kg (28 samples), in the range 1000 – 10000 mg/kg (14 samples) and 10000 – 60000 mg/kg (13 samples), respectively. Sample grouping in two sub-domains was also used in respect with Zn, namely below 1000 mg/kg (30 samples) and over 1000 mg/kg (20 samples). The inspection of narrower ranges of concentration provided a more accurate assessment of results found in HR-CS FAAS and ICP-OES. Plots in Figure 1, Additional file 2 and data in Table 3 demonstrate a good agreement between methods for 95% CI.
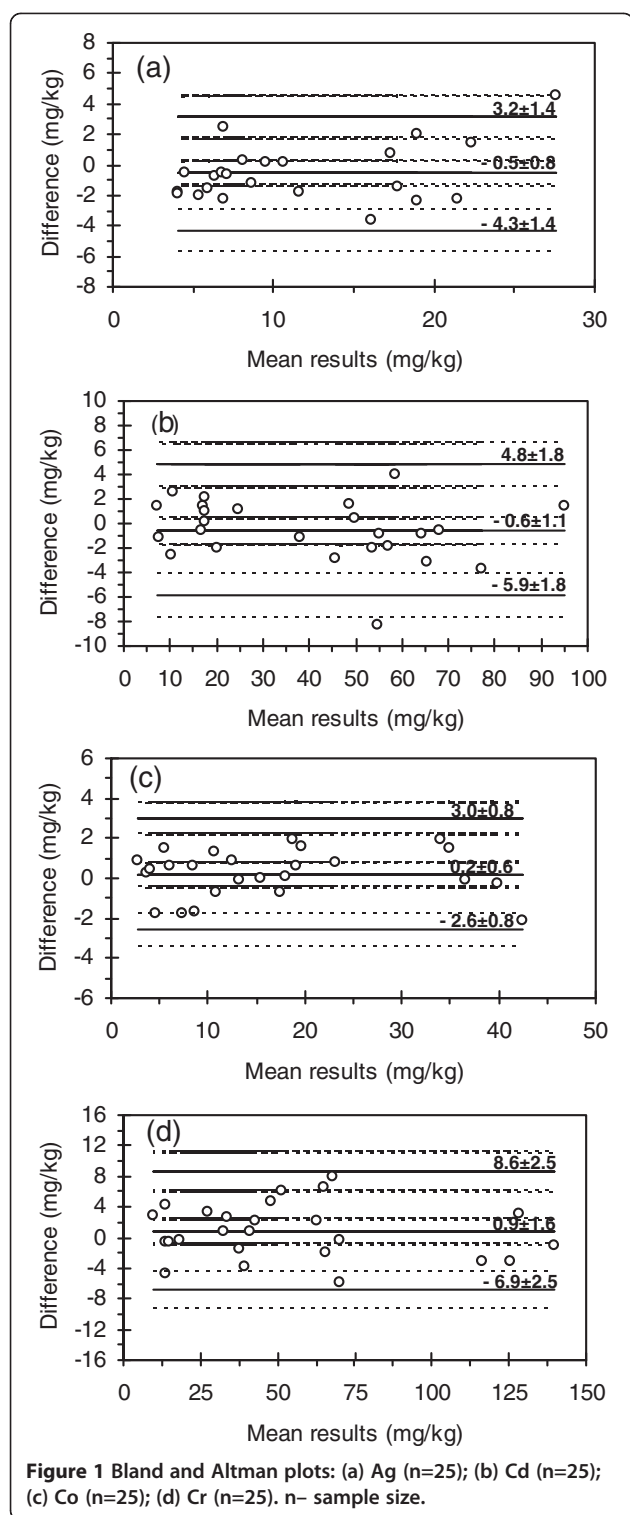
The positive bias for Co and Cr (Figure 1c,d), Cu and Pb < 1000 mg/kg and Zn > 1000 mg/kg (see Additional file 2e,h,l) and the negative one for Ag, Cd (Figure 1a,b), Cu > 1000 mg/kg, Ni, Pb in the range 1000 – 60000 mg/kg and Zn < 1000 mg/kg (see Additional file 2f,g,i,j,k) towards the ICP-OES method was low in relation to the concentrations of the elements. In all these cases the 95% CI of the bias included the zero value, so that errors were random and the null hypothesis was retained. With respect to Co and Cr (Figure 1c,d), Cu (whole concentration range), Ni, Pb (whole concentration range) and Zn (1000 – 11000 mg/kg) (see Additional file 2e,f,g,h,i,j,l), all results fall within the limits of agreement of the differences between methods. In the case of Ag (Figure 1a)

Table 2 Found values (m = 5 independent samples) in the analysis of CRMs compared to certified values (mg/kg)

	BCR 025-050[a]			LGC 6135[a]			LGC 6141[a]			NCSDC			Recovery[b] (%)	
	Found		Certified	Found		Certified	Found		Certified	Found		Certified	ICP-OES	HR-CS AAS
	ICP-OES	HR-CS AAS		ICP-OES	HR-CS AAS		ICP-OES	HR-CS AAS		ICP-OES	HR-CS AAS			
Cd	340 ± 45.0	383 ± 76.0	369 ± 46.3	-		-	-	-	-	2.50 ± 0.5	2.40 ± 0.2	2.45 ± 0.3	97 ± 17	101 ± 15
Co	4.00 ± 0.80	3.98 ± 0.80	4.07 ± 0.93	19 ± 3	16 ± 4	20 ± 4	-	-	-	17.1 ± 1.5	17.0 ± 2.0	16.5 ± 1.8	99 ± 16	94 ± 20
Cr	448 ± 41.0	403 ± 34.0	441 ± 50.1	349 ± 12	344 ± 30	336 ± 28	120 ± 20	115 ± 20	130 ± 31	88 ± 5	105 ± 6	90 ± 8	99 ± 10	100 ± 11
Cu	8.35 ± 1.68	9.62 ± 1.60	7.76 ± 1.68	100 ± 5	102 ± 11	105 ± 5	52.2 ± 0.8	52.3 ± 7.5	51.1 ± 13	54 ± 5	51 ± 2	53 ± 6	102 ± 14	105 ± 12
Ni	12.0 ± 3.25	12.9 ± 1.80	12.2 ± 3.54	240 ± 16	275 ± 27	227 ± 13	50 ± 5	50 ± 5	49 ± 13	34 ± 6	30 ± 3	32	103 ± 17	106 ± 11
Pb	1448 ± 108	1454 ± 188	1447 ± 203	400 ± 21	395 ± 10	391 ± 16	75.1 ± 12	74.1 ± 10	75.8 ± 16	78 ± 5	75 ± 5	79 ± 12	100 ± 10	99 ± 10
Zn	55.0 ± 8.14	60.6 ± 11.05	51.8 ± 8.29	316 ± 18	339 ± 26	316 ± 41	167 ± 30	168 ± 30	169 ± 39	248 ± 38	245 ± 24	251	101 ± 14	105 ± 14

[a] – average ± U (U – expanded uncertainty for 95% confidence level).
[b] – average recovery for 95% confidence level.

Figure 1 Bland and Altman plots: (a) Ag (n=25); (b) Cd (n=25); (c) Co (n=25); (d) Cr (n=25). n– sample size.

between results for Zn by HR-CS FAAS and ICP-OES were random and the null hypothesis was retained over the whole concentration range.

Although in the case of Cd (Figure 1b), one of the 25 results was above the upper limit of agreement and outside its 95% CI, the agreement between HR-CS FAAS and ICP-OES could be considered good.

Conclusions

An analytical method based on the HR-CS FAAS was validated for the fast-sequential determination of eight hazardous/priority hazardous elements (Ag, Cd, Co, Cr, Cu, Ni, Pb and Zn) in soil samples digested in aqua regia. The study revealed similar analytical performances to those in ICP-OES, the standardized method for the multielemental determination in soil and considered as reference in this study. The statistical analysis using the Bland and Altman test demonstrated a good agreement between the two methods over a wide concentration range for 95% CI. In these circumstances it is obvious that the new instrumental concept based on HR-CS FAAS recently commercially available can be successfully used for the determination of hazardous/priority hazardous metals in soil with performances similar to those reported in ICP-OES. The main contribution of the paper was broadening the applicability of HR-CS FAAS for the determination of hazardous/priority hazardous metals in soil, very attractive by its fast, multielemental capability and easy to handle with respect to optimization and overcoming spectral interferences. In the same time, the use of the Bland and Altman test allowed the comparison of results over a large concentration range.

Experimental

Reagents, standards and CRMs

Nitric acid, 65% and hydrochloric acid, 37% of analytical grade (p.a.) (Merck, Darmstadt, Germany) were used to mineralize the soil samples. Argon 5.0 quality and acetylene 4.5 quality (Linde Gas SRL Cluj-Napoca, Romania) were used as working gases to generate ICP and flame, respectively. The accuracy of the HR-CS FAAS method was checked by analyzing three certified reference samples of soil containing priority contaminants, CRM 025–050, LGC 6135, LGC 6141, and one river sediment, NCSDC 78301 (LGC Promochem GmbH, Wessel, Germany). High purity water (18 MΩ cm^{-1}) from a Milli Q system (Millipore, Bedford MA, USA) was used throughout.

The ICP multielement standard solution IV 1000 μg/ml (Merck, Darmstadt, Germany) was used to prepare multielement working standards in the range 0 – 1 μg/ml by dilution with 2% HNO$_3$. A blank calibration solution of 2% HNO$_3$ was used in recording the reference

one result out of 25 was above the upper limit of agreement, while for Zn 2 out of the 20 values > 1000 mg/kg (see Additional file 2,l) surpassed the upper limit and 1 the lower one. However, in all cases the values remained within the 95% CI of the limits. In short, the differences

Table 3 Descriptive statistics for the Bland and Altman test (95% confidence level, m = 3 independent samples)

Element	Within-method standard deviation (mg/kg)		Between-method standard deviation (s_B) (mg/kg)	Confidence interval of bias $\bar{\Delta} \pm t \frac{s_B}{\sqrt{n}}$ (mg/kg)	Confidence intervals for limits of agreement $\bar{\Delta} \pm 1.96 s_B \pm t \cdot s_{LL,UL}$ (mg/kg)
	ICP-OES	HR-CS-AAS			
Ag (n = 25)	0.4	0.3	1.8	−0.5 ± 0.8	−4.1 ± 1.3
					3.1 ± 1.3
Cd (n = 25)	1.0	0.8	2.7	−0.6 ± 1.1	−5.9 ± 1.8
					4.8 ± 1.8
Co (n = 25)	0.7	0.7	1.4	0.2 ± 0.6	−2.6 ± 0.8
					3.0 ± 0.8
Cr (n = 25)	1.3	1.4	3.9	0.9 ± 1.6	−6.9 ± 2.5
					8.6 ± 2.5
Cu (n = 34) (<1000)[a]	30.1	18.7	33.4	4.8 ± 11.7	−60.8 ± 14.0
					70.3 ± 14.0
Cu (n = 16) (1000–22000)[a]	260	246	390	−57 ± 208	−821 ± 258
					707 ± 258
Ni (n = 25)	0.9	1.9	2.5	−0.4 ± 1.0	−5.3 ± 1.3
					4.5 ± 1.3
Pb (n = 28) (<1000)[a]	29.6	22.9	42.5	4.9 ± 16.5	78.3 ± 20.8
					88.2 ± 20.8
Pb (n = 14) (1000–10000)[a]	174	56	214	−36 ± 123	−455 ± 162
					383 ± 162
Pb (n = 13) (10000–60000)[a]	712	494	1477	−243 ± 893	−3138 ± 1343
					2652 ± 1343
Zn (n = 30) (< 1000)[a]	7.5	5.2	13.2	−3.3 ± 4.9	−29.1 ± 6.9
					22.6 ± 6.9
Zn (n = 20) (> 1000)[a]	122	90	169	13 ± 79	−319 ± 100
					345 ± 100

[a] – concentration range in mg/kg; n – sample size.

spectrum for correction NO molecular absorption on Zn 213.856 nm line.

Instrumentation

Flame atomic absorption spectrometer

Measurements in FAAS were carried out using the ContrAA 300 high-resolution continuum source flame atomic absorption spectrometer from Analytik Jena AG (Jena, Germany), currently the only manufacturer of such instrumentation. The spectrometer is equipped with a xenon short-arc lamp XBO 301 (GLE, Berlin, Germany) with a nominal power of 300 W operating in a hot-spot mode suitable for all elements, a double monochromator consisting of a prism pre-monochromator and an echelle grating monochromator (resolution 1.5 pm per pixel in the far UV and 3 pm at 400 nm) and CCD array with 588 pixels. The detector allows visualization of the environment of the analytical line using 200 pixels. Five pixels in the centre of the window were used to measure atomic

absorption, while the others for correction purposes. The concept of ContrAA 300 provides the fast-sequential multielement determination as well as an efficient background correction by simultaneous measurement of atomic and continuous background absorption. Additionally, the interference of fine-structured background of NO molecule on Zn 213.856 nm line was eliminated based on the reference spectrum registered for a 2% HNO_3 solution and the least-squares background correction (LSBC). The primary lines of the elements were selected for the determination due to their highest sensitivity. The working conditions recommended by the manufacturer with the HR-CS ContrAA 300 spectrometer are summarized in Table 4.

Inductively coupled plasma optical emission spectrometer

A SPECTRO CIROS[CCD] ICP optical emission simultaneous spectrometer with axial plasma viewing (Spectro Analytical Instruments, Kleve, Germany) was used to obtain the comparative values for the Bland and Altman

Table 4 Working conditions for measurements by HR-CS FAAS ContrAA 300 in air-acetylene flame

Analyte	Wavelength (nm)	Acetylene (l/min)	Air (l/min)	Burner height (mm)
Ag	328.068	50	500	6
Cd	228.802	50	500	6
Co	240.206	50	500	6
Cr	357.869	100	400	8
Cu	324.754	50	470	6
Ni	232.003	55	470	6
Pb	217.000	65	470	6
Zn	213.856	50	470	6

test. The SPECTRO CIROS features a Paschen-Runge spectrometer mount, consisting of a double-grating optical system with 22 CCD detectors. Details about the operating conditions of the spectrometer and the working wavelengths are given in Table 5. As for HR-CS FAAS the most sensitive wavelengths were selected for emission measurements.

Instrumentation used in sample preparation
A mortar grinder Retsch RM 1000 and a sieve shaker Retsch 200 (Hann, Germany), a Memmert UFE 500 oven (Schwabach, Germany), a high-pressure (100 atm) Berghof MW-S3+ system with control temperature until 210°C (Eningen, Germany) and a Sartorius vacuum filter equipment (Goettingen, Germany) were used for sample preparation and mineralization.

Methods
Soil sample preparation
Mineralization of soil samples was performed according to ISO 11466-1999. Samples were dried for 4 h at

Table 6 Operating conditions for the microwave digestion system

	Step			
	1	2	3	4
Temperature (°C)	180	100	100	100
Ramp time (min)	5	1	1	1
Hold time (min)	25	10	1	1
Power (%)[a]	60	20	10	0

[a] – 100% power corresponds to 1450 W.

$105 \pm 5°C$, then crushed and sieved through a 2-mm sieve to remove stones. The fraction < 2 mm was ground and sieved again through a 250-μm sieve. Amounts of 1.0000 g soil sample prepared in this way were subjected to mineralization with 10 ml aqua regia in closed PTFE containers of the microwave digester according to the protocol given in Table 6. The digest was transferred to a 100 ml volumetric flask and diluted with 2% HNO_3 to mark. After filtration, the solution was kept in polyethylene flask until analysis. Elements were determined by HR-CS FAAS and ICP-OES in 3 independent samples and 3 successive measurements on each sample. Dilutions were carried out so that sample concentration fall within the calibration curve.

Bland and Altman method
In the validation study of the method based on HR-CS FAAS for the determination of elements over a wide concentration range, sometimes up to 3 orders of magnitude (Cu, Pb, Zn), the comparison was made with ICP-OES as reference method using the Bland and Altman approach. It was initially applied in the medical field [56,57] and further used for comparative studies in other domains [58-62]. Unlike the linear regression, the Bland

Table 5 Operating conditions and analytical lines for the SPECTRO CIROS[CCD] ICP optical emission spectrometer

Component	Characteristic
Generator	Free-running 27.12 MHz operated at 1400 W
Plasma torch	Axial viewing, torch positioning (mm) X = − 3.9, Y = + 3.6, Z = + 2.6.
	Argon flow rates:
	Outer gas 12 l/min
	Intermediate gas 0.6 l/min
	Nebulizer gas 1 l/min
Sample introduction system	Four channels peristaltic pump, K2 cross-flow nebulizer, double pass Scott type spray chamber, 2 ml/min sample uptake rate
Polychromator	160 – 600 nm double grating Paschen-Runge mounting, chamber filled with Ar
Detector	22 CCDs
Data processing	Background correction: linear and square two points models, best SNR strategy, integration time 48 s and three successive measurements for each sample.
Analytical wavelengths	Ag 328.068 nm; Cd 214.438 nm; Co 228.615 nm; Cr 267.716 nm; Cu 324.754 nm; Ni 305.082 nm; Pb 220.351 nm; Zn 213.856

and Altman test can be applied for statistical evaluation of experimental data even when they are not uniformly distributed over a concentration range but grouped in several sub-domains, which may be the case of soil samples.

Setting up a Bland and Altman method to compare two data sets implies the following computing: (i) the average of repeated measurements for each sample and each method, and the within-sample standard deviation in each method ($s_{i\ HR\text{-}CS\ FAAS}$ and $s_{i\ ICP\text{-}OES}$); (ii) the within-method standard deviation

$$s_{wHR-CSFAAS} = \sqrt{\frac{\sum s_{iHR-CSFAAS}^2}{n}} \qquad (1)$$

$$s_{wICP-OES} = \sqrt{\frac{\sum s_{iICP-OES}^2}{n}} \qquad (2)$$

where n is the sample size; (iii) the differences between averages of measurements of each sample (Δ_i) performed by the two methods, the bias estimated by the average difference ($\bar{\Delta}$) and its standard deviation (s_Δ); (iv) the between-method standard deviation (s_B) for (m) repeated measurements for each sample in each method

$$s_B = \sqrt{s_\Delta^2 + \left(1 - \frac{1}{m}\right)s_{wHR-CSFAAS}^2 + \left(1 - \frac{1}{m}\right)s_{wICP-OES}^2} \qquad (3)$$

(v) the limits of agreement of the results $\bar{\Delta} \pm 1.96 s_B$ and the 95% CI of $\bar{\Delta}$ ($\bar{\Delta} \pm t\frac{s_B}{\sqrt{n}}$) where n is the sample size, while (t) the Student parameter for (n-1) freedom degrees; (vi) the 95% CI of limits of agreement of the mean difference ($\bar{\Delta} \pm 1.96 s_B \pm t \cdot s_{LL,UL}$) where $s_{LL,\ UL}$ is the standard deviation of the lower and upper limits of agreement;

$$s_{LL,UL} = \sqrt{\frac{s_B^2}{n} + \frac{1.96^2}{2s_{B^2}}\left(\frac{s_\Delta^4}{n-1} + \frac{(m-1)s_{wHR-CSFAAS}^4}{nm^2} + \frac{(m-1)s_{wICP-OES}^4}{nm^2}\right)} \qquad (4)$$

Finally the differences between methods are plotted against the averages of the two methods and the lines corresponding to bias, limits of agreement and their 95% CI respectively, are plotted.

According to the Bland and Altman test, there is no significant bias between methods when the differences of the results fall between the limits of agreement and the CI of the mean difference contains the zero value. Moreover, the 95% CI of the upper and lower limits should be reasonably narrow, which is very likely for a large sample size.

Additional files

Additional file 1: Analytical results (mg/kg) for soil samples analysis by HR-CS FAAS and ICP-OES[a].

Additional file 2: Bland and Altman plots.

Abbreviations
CCD: Charge coupled device; CRM: Certified reference material; HR-CS AAS: High-resolution continuum source atomic absorption spectrometry; HR-CS FAAS: High-resolution continuum source flame atomic absorption spectrometry; HR-CS GFAAS: High-resolution continuum source graphite furnace atomic absorption spectrometry; ICP-OES: Inductively coupled plasma optical emission spectrometry; LOD: Limit of detection; LOQ: Limit of quantification; LR-LS AAS: Low-resolution line source atomic absorption spectrometry; LSBC: Least-squares background correction; OES: Optical emission spectrometry; PTFE: Polytetrafluoroethylene; SNR: Signal-to-noise ratio.

Competing interests
The authors declare that they have no competing interests.

Authors' contributions
TF – designed the study and coordinated the preparation of the manuscript, carried out the analysis by inductively coupled plasma optical emission spectrometry, co-worked on the analysis by high-resolution continuum source flame atomic absorption spectrometry and interpreted the results related to analytical performance; MP – performed the data comparison using the Bland and Altman method; worked on the sample preparation and analysis by optical emission and absorption spectrometry; RH – carried out the computational experiments; co-worked on the sample collection and preparation and helped to draft the manuscript. All authors read and approved the final manuscript.

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References
1. Florek S, Becker-Ross H: **High-resolution spectrometer for atomic spectrometry.** *J Anal At Spectrom* 1995, **10**:145–147.
2. Becker-Ross H, Florek S, Heitmann U, Weisse R: **Influence of the spectral bandwidth of the spectrometer on the sensitivity using continuum source AAS.** *Fresen J Anal Chem* 1996, **355**:300–303.
3. Heitmann U, Schutz M, Becker-Ross H, Florek S: **Measurements of the Zeeman-splitting of analytical lines by means of a continuum source graphite furnace atomic absorption spectrometer with linear charge coupled device array.** *Spectrochim Acta Part B* 1996, **51B**:1095–1105.

4. Welz B, Becker-Ross H, Florek S, Heitmann U: *High-Resolution Continuum, Source AAS - the better way to do atomic absorption spectrometry.* Weinheim, Germany: Wiley-VCH; 2005.

5. L'vov BV: **A continuum source vs. line source on the way toward absolute graphite furnace atomic absorption spectrometry.** *Spectrochim Acta Part B* 1999, **54B**:1637–1646.

6. Welz B, Becker-Ross H, Florek S, Heitmann U, Vale MGR: **High-resolution continuum-source atomic absorption spectrometry – What can we expect?** *J Braz Chem Soc* 2003, **14**:220–229.

7. Becker-Ross H, Florek S, Heitmann U, Huang MD, Okruss M, Radziuk B: **Continuum source atomic absorption spectrometry and detector technology: a historical perspective.** *Spectrochim Acta Part B* 2006, **61B**:1015–1030.

8. Welz B, Borges DLG, Lepri FG, Vale MGR, Heitmann U: **High-resolution continuum source electrothermal atomic absorption spectrometry – an analytical and diagnostic tool for trace analysis.** *Spectrochim Acta PartB* 2007, **62B**:873–883.

9. Welz B, Mores S, Carasek E, Vale MGR, Okruss M, Becker-Ross H: **High-resolution continuum source atomic and molecular absorption spectrometry – a review.** *Appl Spectrosc Rev* 2010, **45**:327–354.

10. Taylor A, Branch S, Day MP, Patriarca M, White M: **Atomic spectrometry update. Clinical and biological materials, foods and beverages.** *J Anal At Spectrom* 2011, **26**:653–692.

11. Resano M, Garcia-Ruiz E: **High-resolution continuum source graphite furnace atomic absorption spectrometry: Is it as it sounds? A critical review.** *Anal Bioanal Chem* 2011, **399**:323–330.

12. Ren T, Zhao LJ, Cao J, Zhong RG: **Determination of Pb, Cd and Cr in food by high resolution continuum source graphite furnace atomic absorption spectrometry.** *Spectrosc Spectr Anal* 2012, **32**:2566–2571.

13. Ren T, Zhao LJ, Zhong RG: **Determination of aluminum in wheat flour food by microwave digestion-high resolution continuum source graphite furnace atomic absorption spectrometry.** *Spectrosc Spectr Anal* 2011, **31**:3388–3391.

14. Nunes AM, Acunha TS, Oreste EQ, Lepri FG, Vieira MA, Curtius AJ, Ribeiro AS: **Determination of Ca, Cu, Fe and Mg in fresh and processed meat treated with tetramethylammonium hydroxide by atomic absorption spectrometry.** *J Braz Chem Soc* 2011, **22**:1850–1857.

15. Atilgan S, Akman S, Baysal A, Bakircioglu Y, Szigeti T, Ovari M, Zaray G: **Monitoring of Pd in airborne particulates by solid sampling high-resolution continuum source electrothermal atomic absorption spectrometry.** *Spectrochim Acta PartB* 2012, **70B**:33–38.

16. Castilho INB, Welz B, Vale MGR, de Andrade JB, Smichowski P, Shaltout AA, Colares L, Carasek E: **Comparison of three different sample preparation procedures for the determination of traffic-related elements in airborne particulate matter collected on glass fiber filters.** *Talanta* 2011, **88**:689–695.

17. Araujo RGO, Rennan GO, Vignola F, Castilho INB, Borges DLG, Welz B, Vale MGR, Smichowski P, Ferreira SLC, Becker-Ross H: **Determination of mercury in airborne particulate matter collected on glass fiber filters using high-resolution continuum source graphite furnace atomic absorption spectrometry and direct solid sampling.** *Spectrochim Acta Part B* 2011, **66B**:378–382.

18. Araujo RGO, Welz B, Castilho INB, Vale MGR, Smichowski P, Ferreira SLC, Becker-Ross H: **Determination of antimony in airborne particulate matter collected on filters using direct solid sampling high-resolution continuum source graphite furnace atomic absorption spectrometry.** *J Anal At Spectrom* 2010, **25**:580–584.

19. dos Santos LMG, Welz B, Araujo RGO, Jacob SDC, Vale MGR, Martens A, Martens IBG, Becker-Ross H: **Simultaneous determination of Cd and Fe in beans and soil of different regions of Brazil using high-resolution continuum source graphite furnace atomic absorption spectrometry and direct solid sampling.** *J Agric Food Chem* 2009, **57**:10089–10094.

20. dos Santos LMG, Araujo RGO, Welz B, Jacob SDC, Vale MGR, Becker-Ross H: **Simultaneous determination of Cd and Fe in grain products using direct solid sampling and high-resolution continuum source graphite furnace atomic absorption spectrometry.** *Talanta* 2008, **78**:577–583.

21. Resano M, Briceno J, Belarra MA: **Direct determination of phosphorus in biological samples using a solid sampling-high resolution-continuum source electrothermal spectrometer: comparison of atomic and molecular absorption spectrometry.** *J Anal At Spectrom* 2009, **24**:1343–1354.

22. Borges AR, Becker EM, Lequeux C, Vale MGR, Ferreira SLC, Welz B: **Method development for the determination of cadmium in fertilizer samples using high-resolution continuum source graphite furnace atomic absorption spectrometry and slurry sampling.** *Spectrochim Acta Part B* 2011, **66B**:529–535.

23. Freire AS, Santelli RE: **Trace elements determination in high salinity petroleum produced formation water by high-resolution continuum source graphite furnace atomic absorption spectrometry after matrix separation using Chelex-100 (R) resin.** *Spectrochim Acta Part B* 2012, **71-72B**:92–97.

24. Quadros DPC, Chaves ES, Lepri FG, Borges DLG, Welz B, Becker-Ross H, Curtius AJ: **Evaluation of Brazilian and Venezuelan crude oil samples by means of the simultaneous determination of Ni and V as their total and non-volatile fractions using high-resolution continuum source graphite furnace atomic absorption spectrometry.** *Energy Fuel* 2010, **24**:5907–5911.

25. Dittert IM, Silva JSA, Araujo RGO, Curtius AJ, Welz B, Becker-Ross H: **Simultaneous determination of cobalt and vanadium in undiluted crude oil using high-resolution continuum source graphite furnace atomic absorption spectrometry.** *J Anal At Spectrom* 2010, **25**:590–595.

26. Dittert IM, Silva JSA, Araujo RGO, Curtius AJ, Welz B, Becker-Ross H: **Direct and simultaneous determination of Cr and Fe in crude oil using high-resolution continuum source graphite furnace atomic absorption spectrometry.** *Spectrochim Acta Part B* 2009, **64B**:537–543.

27. de Campos RC, Correia CLT, Vieira F, Saint Pierre TD, Oliveira AC, Goncalves R: **Direct determination of P in biodiesel by high-resolution continuum source graphite furnace absorption spectrometry.** *Spectrochim Acta Part B* 2011, **66B**:352–355.

28. Aramendia M, Florez MR, Piette M, Vanhaecke F, Resano M: **Al determination in whole blood samples as AlF via high-resolution continuum source graphite furnace molecular absorption spectrometry: potential application to forensic diagnosis of drowning.** *J Anal At Spectrom* 2011, **26**:1964–1973.

29. Nunes LS, Barbosa JTP, Fernandez AP, Lemos VA, dos Santos WNL, Korn MGA, Teixeira LSG: **Multi-element determination of Cu, Fe, Ni and Zn content in vegetable oils samples by high-resolution continuum source atomic absorption spectrometry and microemulsion sample preparation.** *Food Chem* 2011, **127**:780–783.

30. de Oliveira SR, Raposo JL, Neto JAG: **Fast sequential multi-element determination of Ca, Mg, K, Cu, Fe, Mn and Zn for foliar diagnosis using high-resolution continuum source flame atomic absorption spectrometry: feasibility of secondary lines, side pixel registration and least-squares background correction.** *Spectrochim Acta Part B* 2009, **64B**:593–596.

31. Oliveira SR, Neto JAG, Nobrega JA, Jones BT: **Determination of macro- and micronutrients in plant leaves by high-resolution continuum source flame atomic absorption spectrometry combining instrumental and sample preparation strategies.** *Spectrochim Acta Part B* 2010, **65B**:316–320.

32. Brandao GC, Matos GD, Ferreira SLC: **Slurry sampling and high-resolution continuum source flame atomic absorption spectrometry using secondary lines for the determination of Ca and Mg in dairy products.** *Microchem J* 2011, **98**:231–233.

33. Dessuy MB, de Jesus RM, Brandao GC, Ferreira SLC, Vale MGR, Welz B: **Fast sequential determination of antimony and lead in pewter alloys using high resolution-continuum source flame atomic absorption spectrometry.** *Food Addit Contam Part A-Chem* 2013, **30**:202–207.

34. Raposo JL, Oliveira SR, Neto JAG, Nobrega JA, Jones BT: **Determination of silicon in lubrication oil by high-resolution continuum source flame atomic absorption spectrometry using least-square background correction and internal standardization.** *Anal Lett* 2011, **44**:2150–2161.

35. Amorim VR, Neto JAG: **Evaluation of lubricating oil preparation procedures for the determination of Al, Ba, Mo, Si and V by high-resolution continuum source FAAS.** *Anal Sci* 2009, **25**:95–100.

36. Resta V, Filho A, Gomes JA: **Different lubricating oil treatments for the determination of Cu, Cr, Fe, Ni, Sb, Pb and Zn by HR-CS FAAS.** *Anal Lett* 2008, **41**:1555–1570.

37. Raposo JL, de Oliveira SR, Caldas NM, Neto JAG: **Evaluation of alternate lines of Fe for sequential multi-element determination of Cu, Fe, Mn and Zn in soil extracts by high-resolution continuum source flame atomic absorption spectrometry.** *Anal Chim Acta* 2008, **627**:198–202.

38. Huang MD, Becker-Ross H, Florek S, Heitmann U, Okruss M: **Determination of phosphorus by molecular absorption of phophorus monoxide using a**

high-resolution continuum source absorption spectrometer and an air-acetylene flame. *J Anal At Spectrom* 2006, **21**:338–345.

39. Ferreira RB, Oliveira SR, Franzini VP, Virgilio A, Raposo JL, Neto JAG: Evaluation of lines of phosphorus and potassium by high-resolution continuum source flame atomic absorption spectrometry for liquid fertilizer analysis. *At Spectrosc* 2011, **32**:56–61.

40. Welz B, Lepri FG, Araujo RGO, Ferreira SLC, Huang MD, Okruss M, Becker-Ross H: Determination of phosphorus, sulfur and the halogens using high-temperature molecular absorption spectrometry in flames and furnaces - a review. *Anal Chim Acta* 2009, **647**:137–148.

41. Resano M, Florez MR: Direct determination of sulfur in solid samples by means of high-resolution continuum source graphite furnace molecular absorption spectrometry using palladium nanoparticles as chemical modifier. *J Anal At Spectrom* 2012, **27**:401–412.

42. Virgilio A, Raposo JL, Cardoso AA, Nobrega JA, Neto JAG: Determination of total sulfur in agricultural samples by high-resolution continuum source flame molecular absorption spectrometry. *J Agric Food Chem* 2011, **59**:2197–2201.

43. Huang MD, Becker-Ross H, Florek S, Heitmann U, Okruss M: Direct determination of total sulfur in wine using a continuum-source atomic-absorption spectrometer and an air-acetylene flame. *Anal Bioanal Chem* 2005, **382**:1877–1881.

44. Baumbach G, Limburg T, Einax JW: Quantitative determination of sulfur by high-resolution graphite furnace molecular absorption spectrometry. *Microchem J* 2013, **106**:295–299.

45. Bucker S, Acker J: Spectrometric analysis of process etching solutions of the photovoltaic industry – Determination of HNO_3, HF, and H_2SiO_6 using high-resolution continuum source absorption spectrometry of diatomic molecules and atoms. *Talanta* 2012, **94**:335–341.

46. Ozbek N, Akman S: Method development for the determination of fluorine in toothpaste via molecular absorption of aluminum mono fluoride using a high-resolution continuum source nitrous oxide/acetylene flame atomic absorption spectrophotometer. *Talanta* 2012, **94**:246–250.

47. Ozbek N, Akman S: Method development for the determination of fluorine in water samples via the molecular absorption of strontium monofluoride formed in an electrothermal atomizer. *Spectrochim Acta Part B* 2012, **69B**:32–37.

48. Huang MD, Becker-Ross H, Florek S, Heitmann U, Okruss M: High-resolution contuinuum electrothermal absorption spectrometry of AlBr and CaBr for the determination of bromine. *Spectrochim Acta Part B* 2008, **63B**:566–570.

49. Brandao GC, Lima DC, Ferreira SLC: The chemical generation of NO for the determination of nitrite by high-resolution continuum source molecular absorption spectrometry. *Talanta* 2012, **98**:231–235.

50. Massart DL, Vandeginste BGM, Buydens IMC: *Handbook of Chemometrics and Qualimetrics, Part A.* Amsterdam: Elsevier; 1997.

51. Gunzler H: *Accreditation and Quality Assurance in Analytical Chemistry.* Berlin: Springer Verlag; 1994.

52. ISO 22036: *Soil quality. Determination of trace elements in extracts of soil by inductively coupled plasma-atomic emission spectrometry (ICP-AES).* 2008.

53. ISO 11047: *Soil quality - Determination of cadmium, chromium, cobalt, copper, lead, manganese, nickel and zinc - Flame and electrothermal atomic absorption spectrometric methods.* 1999.

54. Kaiser H: Die berechnung der nachweisempfindlichkeit. *Spectrochim Acta* 1947, **3**:40–47.

55. Currie LA: Nomenclature in evaluation of analytical methods including detection and quantification capabilities. *Pure Appl Chem* 1995, **67**:1699–1723.

56. Bland JM, Altman GD: Statistical methods for assessing agreement between two methods of clinical measurement. *Lancet* 1986, **i**:307–310.

57. Bland JM, Altman GD: Measuring agreement in method comparison studies. *Stat Methods Med Res* 1999, **8**:135–160.

58. Lyon TDB, Fell GS, McKey K, Scott RD: Accuracy of multi-element analysis of human tissue obtained at autopsy using inductively coupled plasma mass spectrometry. *J Anal At Spectrom* 1991, **6**:559–564.

59. Cheng ASH, Robson MD, Neubauer S, Selvanayagam J: Irreversible myocardial injury: Assessment with cardiovascular delayed-enhancement MR imaging and comparison of 1.5 and 3.0 T-Initial experience. *Radiology* 2007, **242**:735–742.

60. Frentiu T, Ponta M, Anghel SD, Simon A, Marginean I, Cordos E: Statistical evaluation of Cu, Mn and Zn determinations in biological samples by radiofrequency capacitively coupled plasma atomic emission spectrometry using the Bland and Altman test. *Microchim Acta* 2003, **143**:245–254.

61. Frentiu T, Ponta M, Levei E, Gheorghiu E, Kasler I, Cordos EA: Validation of the Tessier scheme for speciation of metals in soil using the Bland and Altman test. *Chem Pap* 2008, **62**:114–122.

62. Frentiu T, Ponta M, Levei E, Senila M, Ursu M, Cordos E: Interlaboratory study on Cu, Pb and Zn determination in soil by inductively coupled plasma optical emission spectrometry using the Bland and Altman test. *J Optoelectron Adv Mater* 2007, **9**:3505–3515.