# New lanostane-type triterpene acids from wolfiporia extensa 

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#### Abstract

Backgroud: Dried sclerotia of Wolfiporia extensa (Polyporaceae) is used to invigorate the spleen and to tranquilize the mind in Chinese herbal medicine. Lanostane-type triterpene acids were regard as major secondary metabolites from dried sclerotia of W. extensa.

Results: Three new lanostane-type triterpene acids, 3-epi-benzoyloxyl-dehydrotumulosic acid (1), 3-epi-(3'-O-methyl malonyloxy)-dehydrotumulosic acid (2) and 3-epi-(3'-hydroxy-3'-methylglutaryloxyl)-dehydrotumulosic acid (3), were isolated from the sclerotia of $W$. extensa, together with 3 known lanostane derivatives (4-6). Their structures were elucidated on the basis of spectroscopic analysis, including 1D and 2D-NMR techniques. Conclusion: Six lanostane derivatives including three new triterpene acids and three known compounds were reported from the sclerotia of $W$. extensa in this paper.


## Background

Dried sclerotia of Wolfiporia extensa (Polyporaceae), well known as 'Fu-Ling' in China, is used to invigorate the spleen and to tranquilize the mind in Chinese herb medicine [1]. In combination with some other herbs, it shows activities as diuretic, sedative and analgesic [2]. Lanostane-type triterpenes were reported as major secondary metabolites, which are characterized with hydroxyl groups at C-16 position, and with a C-21 carboxylic acid group. A number of lanos-tane-type triterpene acids have been reported from dried sclerotia of $W$. extensa, in which some lanostane derivatives showed activities in the anti-tumor, anti-inflammatory and anti-oxidant activities [3-9]. As part of our continuing research on chemical constituents from Traditional Chinese Medicine (TCM) [10-12], three new lanostane-type triterpene acids, 3-epi-benzoyloxyl-dehydrotumulosic acid (1), 3-epi-(3'-O-methyl malonyloxy)-dehydrotumulosic acid (2) and 3-epi-(3'-hydroxy-3'-methylglutaryloxyl)dehydrotumulosic acid (3) were isolated from the dried sclerotia of W. extensa, together with three known lanostane derivatives (4-6) (Figure 1). Here we report the structure elucidation of the new compounds as follows.

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## Results and discussion

The dried sclerotia of W. extensa were extracted with 95\% ethanol as described in Experimental part. The ethanolic extract was concentrated under reduced pressure to small volume and the solution was fractionated with a HPD-826 macroporous adsorptive resin column eluting with $\mathrm{H}_{2} \mathrm{O}$ and $90 \% \mathrm{EtOH}$. The $90 \% \mathrm{EtOH}$ fraction was concentrated and repeatedly fractionated on reverse-phase ODS, and on silica gel column to obtain six lanostane-type triterpene acids (1-6). Of them, 4-6 were identified as known compounds, dehydropachymic acid (4) [7], dehydrotumulosic acid (5) [13] and 3-epi-dehydrotumulosic acid (6) [13] (Figure 1) by spectroscopic methods and comparison with reported data. Compounds 1-3 were identified as new compounds based on a detailed analysis of NMR as described below (Tables 1 and 2).

Compound 1 was obtained as a colourless crystal in $\mathrm{CH}_{3} \mathrm{OH}$. The molecular formula was determined as $\mathrm{C}_{38} \mathrm{H}_{52} \mathrm{O}_{5}$ from its positive HRESI-MS ( $[\mathrm{M}+\mathrm{H}]^{+}, m / z$ 589.3864 ) and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum. The UV spectrum showed absorption at 234 nm , indicating the presence of a $\Delta^{7,9(11)}$ diene moiety, which was further supported by an absorption band at $1641 \mathrm{~cm}^{-1}$ in the IR spectrum. Strong IR absorption at 3400 and $1710 \mathrm{~cm}^{-1}$ indicated the carboxyl group in $\mathbf{1}$ [13]. The ${ }^{1} \mathrm{H}$-NMR spectrum of $\mathbf{1}$ showed signals from two secondary methyls ( $\delta 0.97$ and 0.99 , each $3 \mathrm{H}, \mathrm{d}, J=6.8 \mathrm{~Hz}$ ), five tertiary methyls ( $\delta 0.92,0.95,1.04$,


Figure 1 Structure of compounds 1-6.
1.06 and 1.48, each $3 \mathrm{H}, \mathrm{s}$ ), two oxygen-bearing methylenes [ $\delta 4.52(1 \mathrm{H}, \mathrm{t}, J=6.8 \mathrm{~Hz})$ and $\delta 5.09(1 \mathrm{H}, \mathrm{br} \mathrm{s})$ ], one terminal methylene group at $\delta 4.84(1 \mathrm{H}, \mathrm{s})$ and $4.97(1 \mathrm{H}, \mathrm{s})$, two olefinic methylenes at $[\delta 5.39(1 \mathrm{H}, \mathrm{d}, J=5.6 \mathrm{~Hz})$ and $\delta$ $5.64(1 \mathrm{H}, \mathrm{br}$ s)], together with signals from typical benzoyl group $[\delta 8.18(2 \mathrm{H}, \mathrm{d}, J=7.2 \mathrm{~Hz}), 7.35(2 \mathrm{H}, \mathrm{d}, J=7.6 \mathrm{~Hz})$, $7.46(1 \mathrm{H}, \mathrm{t}, J=7.4 \mathrm{~Hz})$ ] (Table 1). ${ }^{13} \mathrm{C}-\mathrm{NMR}$ and DEPT spectra of 1 showed signals from 38 carbons, including one carboxyl carbon [ $\delta 178.6$ (C-21)], two carbons from terminal methylene group [ $\delta 107.0(\mathrm{C}-31)$ and 156.1 (C24)], four olefinic carbons [ $\delta 116.7$ (C-11), 120.8 (C-7), 142.9 (C-8) and 146.0 (C-9)], two oxygenated methylenes [ $\delta 79.0$ (C-3) and 76.4 (C-16)], seven methyl carbons [ $\delta$ 17.6 (C-18), 21.9 (C-27), 22.0 (C-26), 22.4 (C-29), 22.7 (C19), 26.6 (C-30) and 28.1 (C-28)], signals from benzoyl group [ $\delta 165.9$ (C-7'), 133.2 (C-4'), 131.4 (C-1'), 129.8 (C$\left.2^{\prime}, 6^{\prime}\right)$, and $128.9\left(\mathrm{C}-3^{\prime}, 5^{\prime}\right)$ ], and signals from other fifteen carbons (see Table 2). The aforementioned NMR features were similar to those of 3-epi-dehydrotumulosic acid (6), except for the existence of an additional set of signals arising from the benzoyl group in $\mathbf{1}$ [13].

The downfield shift at C-3 ( $\delta 79.0$ ) in 1, from ( $\delta 75.1$ ) in 6, suggested that the additional benzoyl group was linked to C-3 position of dehydrotumulosic acid moiety. It was further confirmed by the HMBC experiment which showed correlation between $\mathrm{H}-3$ ( $\delta 5.09$ ) with the signal from C-7' ( $\delta$ 165.9) of the benzoyl groups.
The relative configuration was established by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and the NOESY experiment, in which the $\mathrm{H}-3$ appeared as a broad singlet, the NOESY correlations of $\mathrm{H}-3 \beta$ at ( $\delta 5.09$, $1 \mathrm{H}, \mathrm{br}$ s) with $\mathrm{Me}-29 \beta$ at ( $\delta 0.95,3 \mathrm{H}, \mathrm{s}$ ) revealed the
benzoyl linked the $\alpha$ position of C-3 in compound 1 . On the basis of the above evidence, the structure of $\mathbf{1}$ was elucidated as $3 \alpha$-benzoyl-16 $\alpha$-dihydroxyl-lanost-7, $9(11$ ), 24(31)-trien-21-oic acid, named as 3-epi-benzoyloxyldehydrotumulosic acid.

Compound 2 was obtained as a colourless needle in $\mathrm{CH}_{3} \mathrm{OH}$. Careful comparison of ${ }^{13} \mathrm{C}$-NMR spectra of 1 and 2 indicate that both have a similar lanostane skeleton with different substitution group (details in Table 2). Unlike compound 1 with a benzoyl group, compound 2 showed signals from a malonyl group [ $\delta 41.9\left(-\mathrm{CH}_{2}-\right)$, 166.4 (-COO-) and 167.6 (-COO-)] and a methoxyl group [ $\delta 52.2\left(-\mathrm{OCH}_{3}\right)$ ]. HMBC experiment showed correlations between methoxyl proton ( $\delta 3.63$ ) with $3^{\prime}-\mathrm{C}(\delta$ 166.4, from malonyl group) indicated the methyl malonate group [14]. The HMBC experiment of 2 revealed the correlation between H-3 ( $\delta$ 4.86) and C-1' ( $\delta$ 167.6), indicated the 3 -substitution. Thus, compound 2 was established as $3-\alpha$-methyl-malonyl-16 $\alpha$-dihydroxy-lanost7, 9(11), 24(31)-trien-21-oic acid, named as 3-epi-(3'-Omethyl malonyloxy)-dehydrotumulosic acid.
The ${ }^{13}$ C-NMR spectra of 3 showed signals from a lanostane skeleton similar to those of 1 and 2 (Table 2), except with different substitution groups. Except signals from lanostane skeleton in compound $3,{ }^{1} \mathrm{H}-\mathrm{NMR}$ showed signals at $\left[\delta 3.12\left(1 \mathrm{H}, \mathrm{d}, J=15.2 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right), 3.16(1 \mathrm{H}, \mathrm{d}\right.$, $\left.J=15.2 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right), 3.02\left(1 \mathrm{H}, \mathrm{d}, J=14.4 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 3.08(1 \mathrm{H}$, d, $\left.J=14.4 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right)$ and $1.71\left(3 \mathrm{H}, \mathrm{s},-\mathrm{CH}_{3}\right)$ ] along with ${ }^{13} \mathrm{C}-\mathrm{NMR}$ showed signals [ $\delta 171.4\left(\mathrm{C}-1{ }^{\prime}\right), 46.3\left(\mathrm{C}-2^{\prime}\right), 69.9$ (C-3'), 46.4 (C-4'), 174.6 (C-5'), and $28.4\left(-\mathrm{CH}_{3}\right)$ ]. Those signals were assigned to 3-hydroxy-3-methylglutaryl group based on HMQC and HMBC spectra data. It was further confirmed from ESI-MS experiment, which showed fragment ions at $m / z 525.4\left[\mathrm{M}-\mathrm{H}-102\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)(\mathrm{OH})-\mathrm{CH}_{2}-\right.\right.$ $\mathrm{COOH})]^{-}$. The HMBC correlations of H-3 ( $\delta 4.94 \mathrm{br}$ s) with $\mathrm{C}-1$ ' ( $\delta$ 171.4) confirmed that the 3-hydroxy-3methylglutaryloxyl group was at C-3 in 3 (Figure 1). The compound $\mathbf{3}$ is levorotatory. The $R$-configurations of $\mathrm{C}\left(3^{\prime}\right)$ in $\mathbf{3}$ was deduced by comparing of the compound 3 specific rotation features with those of (+)-3-epi-dehydrotumulosic acid, and ( $3^{\prime} S$ )-(+)-3-hydroxy-3-methylglutaric acid, which are dextrorotatory $[8,13]$. These evidences indicated $R$-configuration of $C\left(3^{\prime}\right)$ in compound 3 . As stated above, the structure of 3 was indicated as $3-\alpha-\left(3^{\prime}\right.$-hydroxy- $3^{\prime}$ -methylglutaryloxy)-16 $\alpha$-dihydroxy-lanost-7, 9(11), 24 (31)-trien-21-oic acid, named as 3-epi-(3'-hydroxy-3'-methylglutaryloxyl)-dehydrotumulosic acid.

## Experimental

## General experimental procedures

Optical rotations were measured on a P-1020 Polarimeter (JASCO, Tokyo, Japan). UV spectra were obtained on a UV 210A Shimadzu spectrometer. IR spectra were recorded on an FT-IR spectrometer (Nicolet iS10, Thermo Scientifi,

Table $1^{1} \mathrm{H}$-NMR data of $\mathbf{1 - 3}$ (at 500 or $\mathbf{6 0 0} \mathbf{~ M H z}$, in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N} ; \delta$ in ppm, Jin Hz )

| position | 1 | 2 | 3 |
| :---: | :---: | :---: | :---: |
| 1 | $\begin{gathered} 1.73, m 1.82, \mathrm{td} \\ (9.6,3.2) \end{gathered}$ | $\begin{aligned} & 1.65, \text { m 1.75, } \\ & \text { dd (13.8, 3.0) } \end{aligned}$ | 1.69, m 1.88, m |
| 2 | 1.91, m 1.97, m | $\begin{aligned} & 1.79, \text { dt (6.6, 3.0) } \\ & \text { 1.85, d (12.0) } \end{aligned}$ | $\begin{gathered} 1.78, \operatorname{ddd}(15.6,6.4 \\ 2.8) 1.86, \mathrm{~m} \end{gathered}$ |
| 3 | 5.09, br s | 4.86, br s | 4.94, br s |
| 5 | $\begin{gathered} 1.88, \mathrm{dd} \\ (10.0,4.0) \end{gathered}$ | 1.68, t (5.1) | 1.76, dd (9.2, 6.4) |
| 6 | 2.08, m 2.09, m | 2.00, m 2.01, m | 2.02, m 2.03, m |
| 7 | 5.64 , br s | 5.57, br s | 5.57, br s |
| 11 | 5.39, d (5.6) | $5.38, \mathrm{~d}$ (6.0) | 5.39, d (6.0) |
| 12 | $\begin{gathered} 2.42, \mathrm{dd} \\ (15.6,5.2) \\ 2.66, \mathrm{~d}(16.8) \end{gathered}$ | $\begin{gathered} \text { 2.42, dd (18.0, 6.6) } \\ 2.66, d(18.0) \end{gathered}$ | $\begin{gathered} \text { 2.42, dd (17.2, 6.8) } \\ 2.66, d(16.4) \end{gathered}$ |
| 15 | $\begin{gathered} 1.95, \mathrm{~d}(12.4) \\ 2.47, \mathrm{dd}(12.8,9.2) \end{gathered}$ | $\begin{aligned} & \text { 1.91, d (13.2) } \\ & 2.45, \mathrm{t}(3.9) \end{aligned}$ | $\begin{aligned} & \text { 1.91, m 2.45, } \\ & \text { dd (12.4, 8.8) } \end{aligned}$ |
| 16 | $4.52, \mathrm{t}$ (6.8) | 4.51, t (7.2) | 4.52, t (6.8) |
| 17 | 2.86 , dd (11.2, 5.6) | 2.85, dd (11.4, 6.0) | 2.84, dd (11.2, 5.6) |
| 18 | 1.06, s | 1.05, s | 1.04, s |
| 19 | 1.04, s | 0.99, s | 1.00, s |
| 20 | 2.95, td (10.8, 2.4) | 2.94, dd (10.8, 3.0) | 2.92, td (10.8, 2.0) |
| 22 | 2.46, m 2.68, m | 2.51, m 2.63, m | 2.42, m 2.61, m |
| 23 | $\begin{gathered} 2.37, \text { m 2.55, } \\ \text { br d (11.6) } \end{gathered}$ | 2.38, m 2.54, m | 2.38, m 2.54, m |
| 25 | 2.29, m | 2.29, m | 2.27, m |
| 26 | 0.97, d (6.8) | 0.97, d (6.6) | 0.97, d (6.8) |
| 27 | 0.99, d (6.8) | 0.98, d (6.6) | 0.99, d (6.8) |
| 28 | 0.92, s | 0.87, s | 0.90, s |
| 29 | 0.95, s | 0.90, s | 0.96, s |
| 30 | 1.48, s | 1.42, s | 1.41, s |
| 31 | $\begin{gathered} \text { 4.84, br s } 4.97, \\ \text { br s } \end{gathered}$ | 4.83, br s 4.97, br s | 4.83 , br s 4.96, br s |
| $2^{\prime}$ | 8.18, d (7.2) | 3.60 , s | $\begin{aligned} & 3.12, d(15.2) \\ & 3.16, d(15.2) \end{aligned}$ |
| 3' | 7.35, t (7.6) | - | - |
| $4^{\prime}$ | 7.46, t (7.4) | $3.63,5$ | $\begin{aligned} & 3.02, \mathrm{~d}(14.4) \\ & 3.08, \mathrm{~d}(14.4) \end{aligned}$ |
| $5^{\prime}$ | 7.35, t (7.6) | - | - |
| $6{ }^{\prime}$ | 8.18, d (7.2) | - | - |
| $-\mathrm{CH}_{3}$ | - | - | 1.71, s |

USA) with KBr pellets. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectrum was recorded in pyridine- $d_{5}$ with Bruker AM-400, DRX-500 and VARIAN INOVA-600 spectrometers operating at 400, 500 and 600 MHz for ${ }^{1} \mathrm{H}$-NMR experiments, and 125 and 150 MHz for ${ }^{13} \mathrm{C}$-NMR experiment, respectively. Coupling constants are expressed in Hertz $(H z)$ and chemical shifts are given on a $\delta$ (ppm) scale with tetramethylsilane as internal standard. Negative ion ESI-MS and HRESI-MS
were recorded on an AutoSpec 3000 spectrometer (VG, Manchester, UK). Column chromatography separations were performed using HPD-826 (Cangzhou Bon Adsorber Technology Co., Cangzhou, China), Chromatorex ODS

Table $2{ }^{13} \mathrm{C}$-NMR Data of $\mathbf{1 - 3}$ (at 125 or 150 MHz , in $\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N} ; \delta$ in ppm)

| Position | 1 | 2 | 3 |
| :---: | :---: | :---: | :---: |
| 1 | 31.2 | 30.8 | 31.1 |
| 2 | 23.5 | 23.2 | 23.4 |
| 3 | 79.0 | 79.6 | 78.2 |
| 4 | 37.7 | 36.8 | 36.7 |
| 5 | 45.3 | 44.7 | 44.8 |
| 6 | 23.2 | 23.1 | 23.1 |
| 7 | 120.8 | 120.8 | 120.7 |
| 8 | 142.9 | 142.7 | 142.8 |
| 9 | 146.0 | 146.0 | 146.0 |
| 10 | 37.2 | 37.6 | 37.6 |
| 11 | 116.7 | 116.6 | 116.5 |
| 12 | 36.2 | 36.2 | 36.2 |
| 13 | 45.1 | 45.1 | 45.1 |
| 14 | 49.5 | 49.5 | 49.5 |
| 15 | 44.4 | 44.4 | 44.4 |
| 16 | 76.4 | 76.4 | 76.4 |
| 17 | 57.6 | 57.6 | 57.6 |
| 18 | 17.6 | 17.6 | 17.6 |
| 19 | 22.7 | 22.6 | 22.7 |
| 20 | 48.5 | 48.5 | 48.5 |
| 21 | 178.6 | 178.7 | 178.6 |
| 22 | 31.4 | 31.4 | 31.4 |
| 23 | 33.2 | 33.2 | 33.2 |
| 24 | 156.1 | 156.0 | 156.1 |
| 25 | 34.1 | 34.1 | 34.1 |
| 26 | 22.0 | 22.0 | 22.0 |
| 27 | 21.9 | 21.8 | 21.8 |
| 28 | 28.1 | 27.9 | 28.1 |
| 29 | 22.4 | 22.3 | 22.5 |
| 30 | 26.6 | 26.6 | 26.6 |
| 31 | 107.0 | 107.0 | 107.2 |
| $1^{\prime}$ | 131.4 | 167.6 | 171.4 |
| $2^{\prime}$ | 129.8 | 41.9 | 46.3 |
| $3^{\prime}$ | 128.9 | 166.4 | 69.9 |
| $4^{\prime}$ | 133.2 | 52.2 | 46.4 |
| $5^{\prime}$ | 128.9 | - | 174.6 |
| $6{ }^{\prime}$ | 129.8 | - | - |
| $7{ }^{\prime}$ | 165.9 | - | - |
| 3'-Me | - | - | 28.4 |

(Fuji Silysia Chemical Co., Greenville, USA) and Silica gel (Qingdao Haiyang Chemical Co., Qingdao, China) as adsorbants. TLC was carried on silica gel $G$ precoated plates (Qingdao Haiyang Chemical Co., Qingdao, China). The TLC plate was monitored by spraying with $10 \%$ $\mathrm{H}_{2} \mathrm{SO}_{4}$ solution in ethanol followed by heating.

## Fungal material

The dried sclerotia of W. extensa were collected from Hebei Guang Ming Prepared Medicinal Herbs Co., Ltd, China and identified by Prof. Yu-Ting Cheng (Beijing University of Chinese Medicines). An authentic sample was kept in School of Chinese Pharmacy, Beijing University of Chinese Medicines.

## Extraction and isolation

The dried sclerotia of W. extensa ( 17.5 kg ) were powdered and extracted with exhaustively $95 \% \mathrm{EtOH}$ under reflux. The EtOH extract was concentrated to the small volume ( 3 L ), and applied on a HPD-826 macroporous adsorptive resin ( $15 \mathrm{Kg}, 18 \mathrm{~cm} \times 150 \mathrm{~cm}$ ), eluting with $\mathrm{H}_{2} \mathrm{O}(60 \mathrm{~L})$ and $90 \% \mathrm{EtOH}(80 \mathrm{~L})$. The $90 \% \mathrm{EtOH}$ fraction was concentrated under reduced pressure, and the residue ( 60 g ) was subjected to column chromatography (CC) on silica gel eluted with $\mathrm{CHCl}_{3} / \mathrm{CH}_{3} \mathrm{OH}(4: 1$ to $1: 1$, 5 L ) to obtain eight fractions ( $\mathrm{Fr} 1-\mathrm{Fr} 8$ ). Fr 1, was further fractionated on silica gel eluted with cyclohexane/ $\mathrm{CHCl}_{3}$ (8:1 and $4: 1$, each 1 L ), and ODS eluted with a step gradient of $\mathrm{H}_{2} \mathrm{O} / \mathrm{MeOH}(1: 0 \rightarrow 0: 1)$, and PTLC (Cyclohexane/ $\mathrm{CHCl}_{3} / \mathrm{HOAc}, 3: 1: 0.1$ ) to give $\mathbf{1}(20 \mathrm{mg}), 2$ $(10 \mathrm{mg})$ and $4(10 \mathrm{mg})$. Fraction 2 was fractionated repeatedly on Silica gel $\left(\mathrm{CHCl}_{3} / \mathrm{EtOAc}, 8: 1\right)$ and ODS $\left(\mathrm{CH}_{3} \mathrm{OH} / \mathrm{H}_{2} \mathrm{O}, \quad 75: 25 \rightarrow 85: 15\right)$, eluted with $\mathrm{CHCl}_{3} /$ $\mathrm{CH}_{3} \mathrm{OH}(50: 1)$, to obtain $5(20 \mathrm{mg})$ and $6(10 \mathrm{mg})$ from Fr 2. Fr 3 was subjected to CC on silica gel $\left(\mathrm{CHCl}_{3} /\right.$ EtOAc, 4:1), and preparative TLC on silica gel $\left(\mathrm{CHCl}_{3} /\right.$ EtOAc/HOAc, 1:1:0.1) to obtain 3 ( 20 mg ).

## 3-epi-benzoyloxyl-dehydrotumulosic acid (1)

Colourless needles; ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (in pyridine- $d_{5}$ ): see Table 1.
${ }^{13} \mathrm{C}$-NMR (in pyridine- $d_{5}$ ): see Table 2. IR $(\mathrm{KBr}) \mathrm{cm}^{-1}$ : 3400, 2928, 1710, 1641, 1279, 1175, 895, 800. UV $\lambda \mathrm{MeOH} \max \mathrm{nm}(\log ): 234$ (4.32). HRESI-MS ( $\mathrm{m} / \mathrm{z}$ ): 589.3864 $[\mathrm{M}+\mathrm{H}]^{+}$, calcd for $\mathrm{C}_{38} \mathrm{H}_{53} \mathrm{O}_{5}, 589.3893$. ESIMS ( $\mathrm{m} / \mathrm{z}$ ) (rel. int.): 587.3 [ $\mathrm{M}-1]^{-}$(100), 417.0 (23), 338.9 (4), 208.8 (13).

## 3-epi-(3'-O-methyl malonyloxy)-dehydrotumulosic acid (2)

 Colourless needles; ${ }^{1} \mathrm{H}$-NMR (in pyridine- $d_{5}$ ): see Table $1 .{ }^{13} \mathrm{C}$-NMR (in pyridine- $d_{5}$ ): see Table 2. IR ( KBr ) $\mathrm{cm}^{-1}$ : 3416, 2960, 1736, 1707, 1641, 1254, 1152, 891, 800. UV $\lambda \mathrm{MeOH} \max \mathrm{nm}$ (log): 243 (4.16). HRESI-MS $(m / z): 607.3605[\mathrm{M}+\mathrm{Na}]^{+}$, calcd for $\mathrm{C}_{35} \mathrm{H}_{52} \mathrm{O}_{7} \mathrm{Na}$, 607.3611.3-epi-(3' -hydroxy-3'-methylglutaryloxyl)-dehydrotumulosic acid (3)
Colourless needles; $[\alpha]=-7.6\left(c=0.1705\right.$, pyridine); ${ }^{1} \mathrm{H}-$ NMR (in pyridine- $d_{5}$ ): see Table $1 .{ }^{13} \mathrm{C}$-NMR (in pyri-dine- $d_{5}$ ): see Table 2. IR ( KBr ) $\mathrm{cm}^{-1}: 3389,2962,1707$, 1642, 1205, 1176, 891, 802, 780, 770. UV $\lambda \mathrm{MeOH} \max$ nm (log): 244 (4.13); HRESI-MS ( $m / z$ ): $651.3880[\mathrm{M} \mathrm{+}$ $\mathrm{Na}]^{+}$, calcd for $\mathrm{C}_{37} \mathrm{H}_{56} \mathrm{O}_{8} \mathrm{Na}$, 651.3873. ESI-MS ( $\mathrm{m} / \mathrm{z}$ ) (rel. int.): $627.5\left[\mathrm{M} \mathrm{-} \mathrm{1]}{ }^{-}\right.$(100), 525.4 (5).

## Competing interests

The authors declare that they have no competing interests.

## Authors' contributions

GS carried out the chemical analysis-structure elucidation and drafted the Manuscript; NZ carried out the chemical studies; SW employed in the several chemical assays of extraction and isolation; YL worked at the part of experimental design; YB engaged in the part of chemical analysis-structure elucidation; CS carried out the part of chemical assays of extraction and isolation; SR conceived of the study and its design and coordination of the scientific teams. All authors have read and approved the final manuscript.

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